## HYDRATE ALKALINITY TEST KIT



## **DESCRIPTION**

The HYDRATE ALKALINITY Test Kit contains everything needed to measure boiler water hydrate alkalinity. The presence of hydrate alkalinity in boiler water ensures that any calcium hardness in the presence of phosphate or magnesium hardness in the presence of silica will be in a form that is easily removed by blowdown.

The easy drop count titration method provides accurate results. After barium chloride is added to remove interference from phosphate and carbonate, the sample is then titrated to a phenolphthalein endpoint with sulfuric acid. Results are expressed as ppm OH.

One kit provides approximately 200 tests when the hydrate (hydroxide) alkalinity is 40-65 ppm.

## **APPLICATION & USE**

For low-pressure boiler systems, Drew Marine recommends testing boiler water hydrate alkalinity once a day per boiler.

Before testing, boiler water samples must be cooled to 25° C (77° F) by collecting through a sample cooler for safety and to prevent flashing, which will concentrate the sample.

See reverse side for test procedure.



HYDRATE ALKALINITY TEST KIT (PCN 0388018)

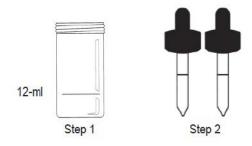
## **TEST KIT CONTENTS**

7 x 60 ml Barium Chloride
3 x 60 ml Sulfuric Acid
30 ml Phenolphthalein
1 Plastic Titration Vial marked at 12-ml
1 Dropper Pipette marked at 0.5 and 1-ml



Ensure proper PPE is worn including latex gloves and safety glasses prior to performing this test. **PROCEDURE** 

- 1. Rinse and fill the plastic titration vial to the line (12-ml) with cooled boiler water sample (Step 1).
- 2. Pipette 2 ml of barium chloride 10% into the vial and swirl to mix (Step 2).
- 3. Add 2 drops of Phenolphthalein indicator and swirl. IF THE SAMPLE DOES NOT TURN PINK, the hydrate alkalinity level is zero. Record zero and adjust treatment dosage to increase hydrate alkalinity. IF THE SAMPLE TURNS PINK, counting the drops, add sulfuric acid until the sample is colorless (disregard the eventual reappearance of a pink color). Swirl the vial between drops (Step 3).
- 4. Calculate the hydrate alkalinity as follows: Number of drops of sulfuric acid x 5 = ppm hydrate alkalinity as OH.
- 5. Record the hydrate alkalinity result and adjust treatment dosage as necessary.







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